

Chiral dopants

Patent Number: ☐ US6217792
Publication date: 2001-04-17
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Requested Patent: ☐ WO9800428
Application Number: US19980214387 19981230
Priority Number (s): EP19960110578 19960701; WO1997EP03167 19970618
IPC Classification: C09K19/34; C09K19/00
EC Classification: C07D493/04, C09K19/58B2
Equivalents: CN1103776B, CN1223657, CN1420382, DE19781752T, ☐ GB2329636, JP2000515496T, KR2000022497

Abstract

The invention relates to chiral dopants of the formula (I): R1-X1-MG1-G-MG2-X2-R2. The invention also relates to liquid crystalline materials comprising at least one chiral dopant of formula (I) and optionally at least one polymerizable mesogenic compound. The invention furthermore relates to the use of such liquid crystalline materials for the preparation of polymer films with a chiral liquid crystalline phase, for active and passive optical elements or color filters and for liquid crystal displays, for example STN, TN, AMD-TN, temperature compensation, guest-host or phase change displays, or polymer free or polymer stabilized cholesteric texture (PFCT, PSCT) displays. The invention also relates to cholesteric liquid crystal displays comprising liquid crystalline materials comprising chiral dopants of formula (I) and to polymer films with a chiral liquid crystalline phase obtainable by (co)polymerizing a liquid crystalline material comprising at least one chiral of formula (I) and at least one polymerizable compound

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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : C07D 493/04, C09K 19/58	A1	(11) International Publication Number: WO 98/00428 (43) International Publication Date: 8 January 1998 (08.01.98)
(21) International Application Number: PCT/EP97/03167 (22) International Filing Date: 18 June 1997 (18.06.97) (30) Priority Data: 96110578.0 1 July 1996 (01.07.96) EP (34) Countries for which the regional or international application was filed: DE et al. (71) Applicant (for all designated States except US): MERCK PATENT GMBH [DE/DE]; Frankfurter Strasse 250, D-64293 Darmstadt (DE). (72) Inventors; and (75) Inventors/Applicants (for US only): PARRI, Owain [GB/GB]; 45 Houlton Road, Poole, Dorset BH15 2LN (GB). NOLAN, Patrick [GB/GB]; 31 Houlton Road, Poole, Dorset BH15 2LN (GB). FARRAND, Louise [GB/GB]; 12 Cinnamon Lane, Poole, Dorset BH15 1JL (GB). MAY, Alison [GB/GB]; 10 Hadrian Way, Corfe Mullen, Wimborne, Dorset BH21 3XF (GB). (74) Common Representative: MERCK PATENT GMBH; Frankfurter Strasse 250, D-64293 Darmstadt (DE).		(81) Designated States: CN, DE, GB, JP, KR, US. Published <i>With international search report.</i>
(54) Title: CHIRAL DOPANTS		
(57) Abstract The invention relates to chiral dopants of the formula (I): $R^1-X^1-MG^1-G-MG^2-X^2-R^2$. The invention also relates to liquid crystalline materials comprising at least one chiral dopant of formula (I) and optionally at least one polymerizable mesogenic compound. The invention furthermore relates to the use of such liquid crystalline materials for the preparation of polymer films with a chiral liquid crystalline phase, for active and passive optical elements or colour filters and for liquid crystal displays, for example STN, TN, AMD-TN, temperature compensation, guest-host or phase change displays, or polymer free or polymer stabilized cholesteric texture (PFCT, PSCT) displays. The invention also relates to cholesteric liquid crystal displays comprising liquid crystalline materials comprising chiral dopants of formula (I) and to polymer films with a chiral liquid crystalline phase obtainable by (co)polymerizing a liquid crystalline material comprising at least one chiral of formula (I) and at least one polymerizable compound.		

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Chiral D pants

The invention relates to chiral dopants of formula I

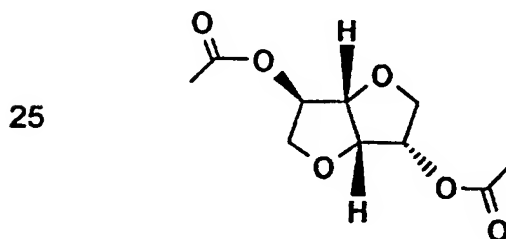


wherein

10 R^1 and R^2 are independently of each other a straight-chain or branched alkyl radical with up to 25 C atoms which may be unsubstituted, mono- or polysubstituted by halogen or CN, it being also possible for one or more non-adjacent CH_2 groups to be replaced, in each case independently from one another, by -
 15 $O-$, $-S-$, $-NH-$, $-N(CH_3)-$, $-CO-$, $-COO-$, $-OCO-$, $-OCO-O-$, $-S-CO-$, $-CO-S-$ or $-C\equiv C-$ in such a manner that oxygen atoms are not linked directly to one another,

20 X^1 and X^2 are independently of each other $-O-$, $-S-$, $-CO-$, $-COO-$, $-OCO-$, $-OCO-O-$, $-S-CO-$, $-CO-S-$ or a single bond,

G is the following chiral bivalent structure element



and

30 MG^1 and MG^2 are each independently a mesogenic or mesogeneity supporting group of formula II



35 with

Z denoting -COO-, -OCO-, -CH₂CH₂-, -OCH₂-, -CH₂O-, -CH=CH-, -CH=CH-COO-, -OCO-CH=CH-, -C≡C- or a single bond,

5 A¹ and A² being each independently 1,4-phenylene in which, in addition, one or more CH groups may be replaced by N, 1,4-cyclohexylene in which, in addition, one or two non-adjacent CH₂ groups may be replaced by O and/or S, 1,4-cyclohexenylene, 1,4-bicyclo(2,2,2)octylene, piperidine-
10 1,4-diyl, naphthalene-2,6-diyl, decahydronaphthalene-2,6-diyl, or 1,2,3,4-tetrahydronaphthalene-2,6-diyl, it being possible for all these groups to be unsubstituted, mono- or polysubstituted with halogen, cyano or nitro groups or alkyl, alkoxy or alkanoyl groups having 1 to 7 C atoms wherein
15 one or more H atoms may be substituted by F or Cl, and

m being 0, 1, 2 or 3.

20 The invention also relates to a liquid crystalline material comprising at least one chiral dopant of formula I and optionally at least one polymerizable mesogenic compound.

25 The invention furthermore relates to the use of such a liquid crystalline material for the preparation of polymer films with a chiral liquid crystalline phase, for active and passive optical elements or colour filters and for liquid crystal displays, for example STN, TN, AMD-TN, temperature compensation, guest-host or phase change displays, or polymer free or polymer stabilized cholesteric texture (PFCT, PSCT) displays.

30 The invention also relates to liquid crystal displays comprising a liquid crystalline material comprising at least one chiral dopant of formula I and to polymer films with a chiral liquid crystalline phase obtainable by (co)polymerizing a liquid crystalline material
35 comprising at least one chiral dopant of formula I and at least one polymerizable mesogenic compound.

Chiral dopants are used to induce or enhance a helical twist of the molecules of a liquid crystalline medium that can be used for example in liquid crystal displays. The pitch p of the molecular helix in the first approximation, which is sufficient for most practical applications, is inverse proportional to the concentration c of the chiral dopant in the liquid crystal host mixture according to equation (1) :

$$p = \frac{1}{\text{HTP}} \cdot \frac{1}{c} \quad (1)$$

The proportionality factor is the helical twisting power (HTP) of the chiral dopant.

- For many applications it is desirable to have LC mixtures that exhibit a twist. Among these are e.g. phase-change displays, guest-host displays, passive and active matrix TN and STN displays like AMD-TN, including such displays with temperature compensated characteristics, e.g. by appropriate selection of the cholesteric dopants according to the invention either alone or in combination with further chiral dopants. For these applications it is advantageous to have available a chiral dopant with a high HTP in order to reduce the amount of dopant needed to induce the desired pitch.
- For some applications it is desired to have LC mixtures that exhibit a strong helical twist and thereby a short pitch length. For example in liquid crystalline mixtures that are used in selectively reflecting cholesteric displays, the pitch has to be selected such that the maximum of the wavelength reflected by the cholesteric helix is in the range of visible light. Another possible application are polymer films with a chiral liquid crystalline phase for optical elements, such as cholesteric broadband polarizers or chiral liquid crystalline retardation films.
- As can be seen from equation (1), a short pitch can be achieved by using high amounts of dopant or by using a dopant with a high HTP.

5 However, the chiral dopants of prior art often exhibit low values of the HTP, so that high amounts of dopant are needed. This is a disadvantage because, as chiral dopants can be used only as pure enantiomers, they are expensive and difficult to synthesize.

10 Further and in many cases even more important, when using chiral dopants of prior art in high amounts, they often negatively affect the properties of the liquid crystalline host mixture, such as e.g. the dielectric anisotropy $\Delta\epsilon$, the viscosity, the driving voltage or the switching times.

15 There is thus a considerable demand for chiral dopants with a high HTP which are easy to synthesize, which can be used in low amounts, show improved temperature stability of the cholesteric pitch e.g. for utilizing a constant reflection wavelength and do not affect the properties of the liquid crystalline host mixture.

20 The invention has the object of providing chiral dopants having these properties, but which do not have the disadvantages of the dopants of the state of the art as discussed above.

25 It has been found that this object can be achieved by the provision of chiral dopants according to formula I.

The inventive chiral dopants contain a chiral structure element based on 1,4:3,6-Dianhydro-D-sorbitol, which is economically and easily available from sugars as natural source.

30 Chiral polymerizable mesogenic compounds comprising 1,4:3,6-Dianhydro-D-sorbitol as structure element are disclosed in WO 95/16007.

35 G. Wulff et al. in Makromolekulare Chemie, 188 (4), 731-40 (1987) describe non-mesogenic 1,4:3,6-Dianhydro-D-sorbitol-2,5-

dimethacrylate used as chiral crosslinking agent for template-imprinted vinyl and acrylic polymers.

5 However the compounds described in WO 95/16007 and by G. Wulff et al. are reactive and consequently not sufficiently stable for most applications.

Thus the object of this invention are chiral dopants of formula I

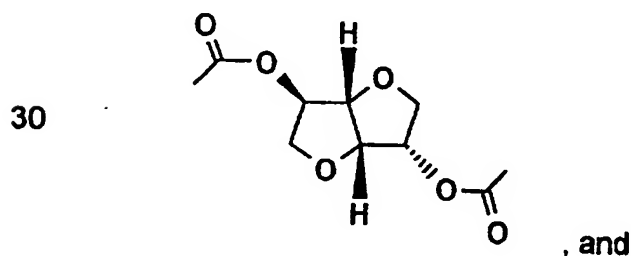
10 $R^1-X^1-MG^1-G-MG^2-X^2-R^2$ I

wherein

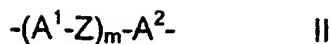
15 R^1 and R^2 are each independently a straight-chain or branched alkyl radical with up to 25 C atoms which may be unsubstituted, mono- or polysubstituted by halogen or CN, it being also possible for one or more non-adjacent CH_2 groups to be replaced, in each case independently from one another, by -O-,
 20 -S-, -NH-, -N(CH₃)-, -CO-, -COO-, -OCO-, -OCO-O-, -S-CO-, -CO-S- or -C≡C- in such a manner that oxygen atoms are not linked directly to one another,

25 X^1 and X^2 are independently of each other -O-, -S-, -CO-, -COO-, -OCO-, -OCO-O-, -S-CO-, -CO-S- or a single bond,

G is the following chiral bivalent structure element



35 MG^1 and MG^2 are each independently a mesogenic or mesogeneity supporting group of the formula II



with

5

Z denoting -COO-, -OCO-, -CH₂CH₂-, -OCH₂-, -CH₂O-,
-CH=CH-, -CH=CH-COO-, -OCO-CH=CH-, -C≡C- or a
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10

A¹ and A² being each independently 1,4-phenylene in which, in
addition, one or more CH groups may be replaced by N,
1,4-cyclohexylene in which, in addition, one or two non-
adjacent CH₂ groups may be replaced by O and/or S, 1,4-
cyclohexenylene, 1,4-bicyclo(2,2,2)octylene, piperidine-
1,4-diyl, naphthalene-2,6-diyl, decahydronaphthalene-2,6-
15 diyl, or 1,2,3,4-tetrahydronaphthalene-2,6-diyl, it being
possible for all these groups to be unsubstituted, mono- or
polysubstituted with halogen, cyano or nitro groups or alkyl,
alkoxy or alkanoyl groups having 1 to 7 C atoms wherein
20 one or more H atoms may be substituted by F or Cl, and

20

m being 0, 1, 2 or 3.

25 Another object of the invention is a liquid crystalline material
comprising at least one chiral dopant of formula I and optionally at
least one polymerizable mesogenic compound.

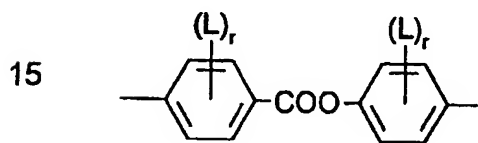
A further object of the invention is the use of a liquid crystalline
material as described above for the preparation of polymer films with
30 a chiral liquid crystalline phase, for active and passive optical
elements, colour filters and for liquid crystal displays, for example
STN, TN, AMD-TN, temperature compensation, guest-host or phase
change displays, or polymer free or polymer stabilized texture
(PFCT, PSCT) cholesteric displays .

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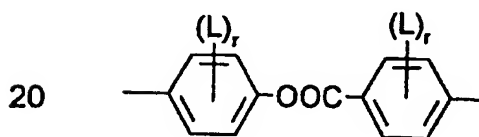
Another object of the invention are cholesteric liquid crystal displays comprising a liquid crystalline material comprising at least one chiral dopant of formula I.

5 Yet another object of the invention are polymer films with a chiral liquid crystalline phase obtainable by (co)polymerizing a liquid crystalline material comprising at least one chiral dopant of formula I and at least one polymerizable mesogenic compound.

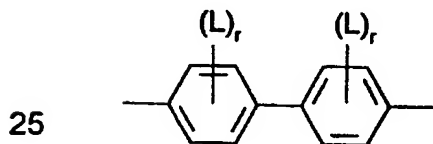
10 Of the chiral dopants of formula I particularly preferred are those wherein MG^1 and MG^2 are selected according to the following formulae:



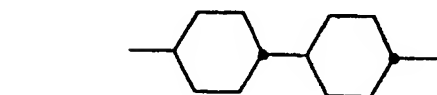
II a



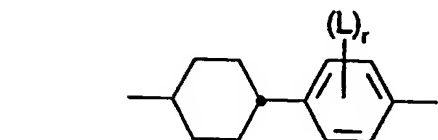
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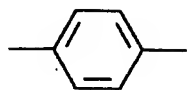
II c



II d

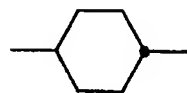


II e



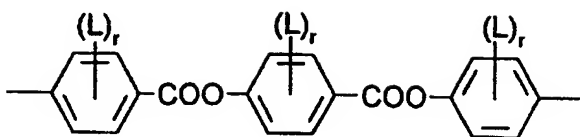
II f

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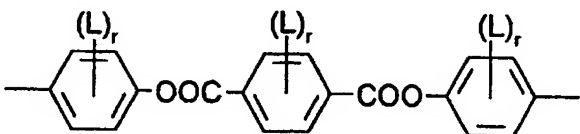
II g

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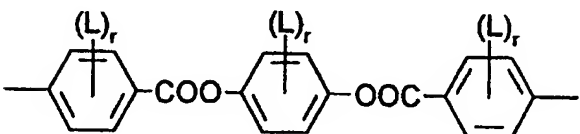
II h

15



II i

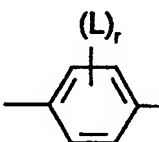
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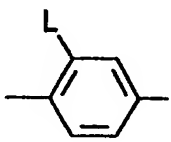
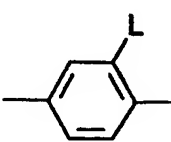
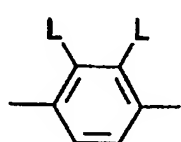
II k

In the formulae II a to II k L denotes, in each case, independently from each other, halogen, a cyano or nitro group or an alkyl, alkoxy or alkanoyl group having 1 to 7 C atoms wherein one or more H atoms may be substituted by F or Cl, and r is 0, 1 or 2.

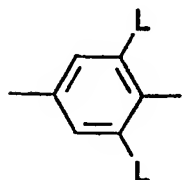
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The group  in these preferred formulae is very preferably

30

denoting  ,  or  , furthermore

35



, with L having each independently one of the meanings given above.

From the preferred compounds those wherein MG^1 and MG^2 are selected of formula II a, II b, II c, II d and II e are particularly preferred.

- 5 From the preferred compounds especially preferred are those wherein MG^1 and MG^2 are selected of formula II a and II b.

10 In these especially preferred compounds r is particularly preferably 0. Furthermore preferred are compounds with MG^1 and MG^2 being of formula II a and II b wherein at least two aromatic rings are substituted by (L)_r with r being 1 or at least one ring is substituted by (L)_r with r being 2, with L having the meaning given above.

15 If R^1 and/or R^2 are an alkyl or alkoxy radical, i.e. where the terminal CH_2 group is replaced by -O-, this may be straight-chain or branched. It is preferably straight-chain, has 2, 3, 4, 5, 6, 7 or 8 carbon atoms and accordingly is preferably ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, ethoxy, propoxy, butoxy, pentoxy, hexoxy, heptoxy, or octoxy, furthermore methyl, nonyl, decyl, undecyl, dodecyl, tridecyl, 20 tetradecyl, pentadecyl, methoxy, nonoxy, decoxy, undecoxy, dodecoxy, tridecoxy or tetradecoxy, for example.

25 Oxaalkyl, i.e. where one CH_2 group is replaced by -O-, is preferably straight-chain 2-oxapropyl (=methoxymethyl), 2- (=ethoxymethyl) or 3-oxabutyl (=2-methoxyethyl), 2-, 3-, or 4-oxapentyl, 2-, 3-, 4-, or 5-oxahexyl, 2-, 3-, 4-, 5-, or 6-oxaheptyl, 2-, 3-, 4-, 5-, 6- or 7-oxaoctyl, 2-, 3-, 4-, 5-, 6-, 7- or 8-oxanonyl or 2-, 3-, 4-, 5-, 6-, 7-, 8- or 9-oxadecyl, for example.

30 R^1 and R^2 in the preferred compounds are preferably alkyl or alkoxy with 1 to 12 C atoms.

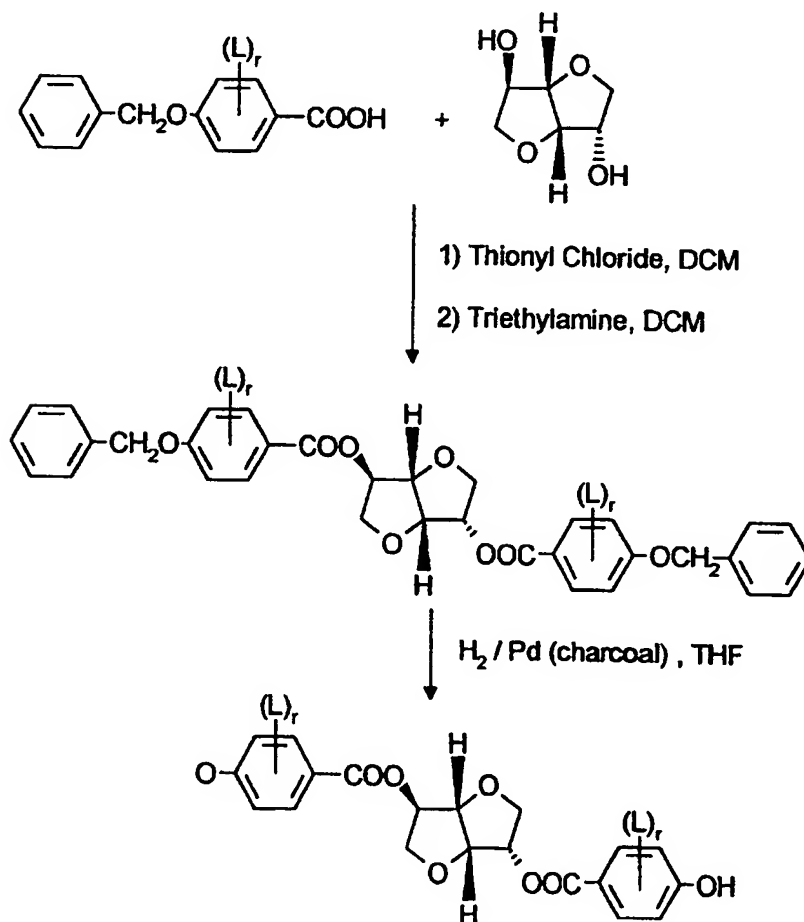
35 X^1 and X^2 in formula I are preferably denoting -O-, -CO-, -COO-, -OCO- or a single bond.

L is preferably F, Cl, CN, NO₂, CH₃, C₂H₅, OCH₃, OC₂H₅, COCH₃, COC₂H₅, CF₃, OCF₃, OCHF₂, OC₂F₅, in particular F, Cl, CN, CH₃, C₂H₅, OCH₃, COCH₃ and OCF₃, most preferably F, CH₃, OCH₃ and COCH₃.

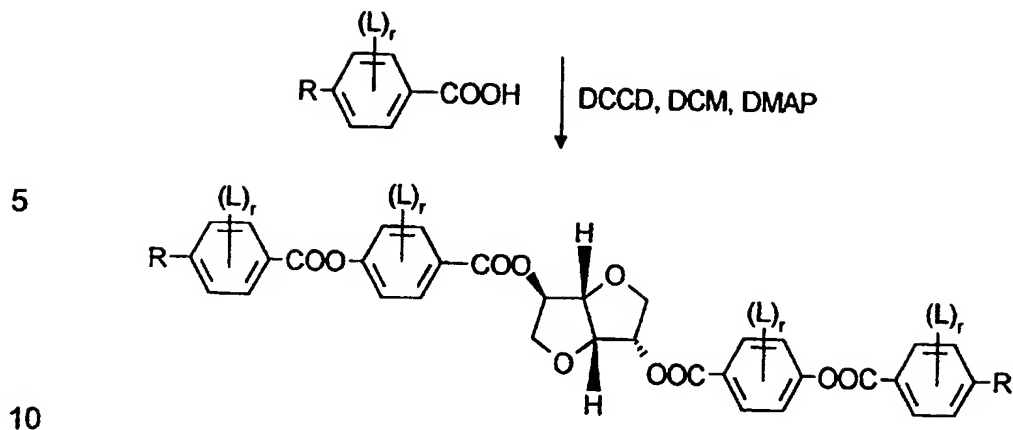
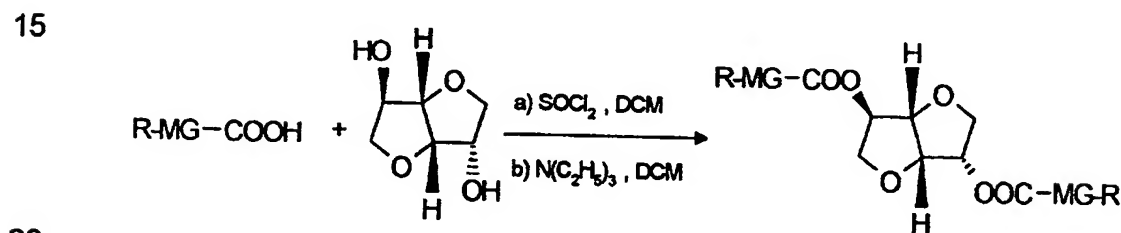
In the compounds of formula I R¹ and R² as well as MG¹ and MG² on both sides of the bivalent chiral group G can be identical or different. Particularly preferred are the compounds of formula I wherein R¹ and R² as well as MG¹ and MG² are identical.

The inventive chiral dopants can be synthesized according to or in analogy to the following reaction schemes:

Scheme 1:



- 11 -

Scheme 2

DCCD = dicyclohexylcarbodiimide, DMAP = dimethylaminopyridine,
DCM = dichloromethane.

25 In the reaction schemes 1 and 2 R is denoting an alkyl or alkoxy
group with 1 to 12 C atoms, L and r have the meanings given above
and MG has one of the meanings of formula II a to II g.

30 The inventive chiral dopants can be used in liquid crystal materials
for displays exhibiting a twisted molecular structure of the liquid
crystal matrix like, for example, supertwisted or active matrix liquid
crystal displays, or in displays comprising a liquid crystal material
with a chiral liquid crystalline phase, like for example chiral smectic
or chiral nematic (cholesteric) materials for ferroelectric displays or
cholesteric displays.

35

Thus, another object of the invention is a liquid crystalline material comprising at least one chiral dopant of formula I.

5 Yet another object of the invention are cholesteric liquid crystal displays comprising cholesteric liquid crystalline materials containing at least one chiral dopant of formula I.

10 The inventive chiral dopants of formula I exhibit high values of the HTP. Thus liquid crystalline materials with a high helical twist, i.e. a short cholesteric pitch, can be prepared by using the inventive dopants, or otherwise liquid crystalline materials with a medium helical twist can be achieved already when using the inventive dopants in low amounts.

15 The high HTP values of the inventive dopants makes them also suitable to be used in combination with other dopants for the temperature compensation of the properties of liquid crystal mixtures, such as the cholesteric pitch, and of the properties of displays, e.g. such as the threshold voltage.

20 In a preferred embodiment of the invention the chiral dopants show a strong temperature dependence of the HTP in nematic liquid crystal mixtures.

25 The inventive dopants are furthermore advantageous because they are affecting the physical properties of the liquid crystalline material only to a minor extent.

30 Thus, when admixing the chiral dopants of formula I for example to a liquid crystalline material with positive dielectric anisotropy that is used in a liquid crystal display, $\Delta\epsilon$ is being only slightly reduced and the viscosity of the liquid crystalline material is increased only to a small extent. This leads to lower voltages and improved switching times of the display when compared to a display comprising
35 conventional dopants.

In a particularly preferred embodiment of the invention the chiral dopants show a small temperature dependence of the HTP in nematic liquid crystal mixtures.

5 The liquid crystalline material according to the invention comprises preferably 0.001 to 15 %, in particular 0.01 to 7 % and very particularly preferably 0.1 to 4 % by weight of chiral dopants of formula I.

10 The liquid crystalline material according to the invention preferably comprises 1 to 3, very preferably 1 or 2, in particular 1 chiral dopants of formula I.

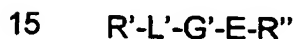
15 For temperature compensation applications as described above the liquid crystalline material preferably contains a chiral component which contains at least one chiral dopant of formula I.

20 In a preferred embodiment of the invention the liquid crystalline material is consisting of 2 to 25, preferably 3 to 15 compounds, at least one of which is a chiral dopant of formula I. The other
25 compounds are preferably low molecular weight liquid crystalline compounds selected from nematic or nematogenic substances, for example from the known classes of the azoxybenzenes, benzylideneanilines, biphenyls, terphenyls, phenyl or cyclohexyl benzoates, phenyl or cyclohexyl esters of cyclohexanecarboxylic acid, phenyl or cyclohexyl esters of cyclohexylbenzoic acid, phenyl or cyclohexyl esters of cyclohexylcyclohexanecarboxylic acid, cyclohexylphenyl esters of benzoic acid, of cyclohexanecarboxylic acid and of cyclohexylcyclohexanecarboxylic acid,
30 phenylcyclohexanes, cyclohexylbiphenyls, phenylcyclohexylcyclohexanes, cyclohexylcyclohexanes, cyclohexylcyclohexenes, cyclohexylcyclohexylcyclohexenes, 1,4-bis-cyclohexylbenzenes, 4,4'-bis-cyclohexylbiphenyls, phenyl- or cyclohexylpyrimidines, phenyl- or cyclohexylpyridines, phenyl- or cyclohexylpyridazines, phenyl- or cyclohexyldioxanes, phenyl- or
35 cyclohexyl-1,3-dithianes, 1,2-diphenylethanes, 1,2-dicyclohexylethanes, 1-phenyl-2-cyclohexylethanes, 1-cyclohexyl-2-

(4-phenylcyclohexyl)ethanes, 1-cyclohexyl-2-biphenylethanes, 1-phenyl-2-cyclohexylphenylethanes, optionally halogenated stilbenes, benzyl phenyl ether, tolans, substituted cinnamic acids and further classes of nematic or nematogenic substances. The 1,4-phenylene groups in these compounds may also be laterally mono- or difluorinated.

The liquid crystalline material of this preferred embodiment is based on the achiral compounds of this type.

The most important compounds that are possible as components of these liquid crystalline materials can be characterized by the following formula



wherein L' and E, which may be identical or different, are in each case, independently from one another, a bivalent radical from the group formed by -Phe-, -Cyc-, -Phe-Phe-, -Phe-Cyc-, -Cyc-Cyc-, -Pyr-, -Dio-, -B-Phe- and -B-Cyc- and their mirror images, where Phe is unsubstituted or fluorine-substituted 1,4-phenylene, Cyc is trans-1,4-cyclohexylene or 1,4-cyclohexenylene, Pyr is pyrimidine-2,5-diyl or pyridine-2,5-diyl, Dio is 1,3-dioxane-2,5-diyl and B is 2-(trans-1,4-cyclohexyl)ethyl, pyrimidine-2,5-diyl, pyridine-2,5-diyl or 1,3-dioxane-2,5-diyl.

G' in these compounds is selected from the following bivalent groups -CH=CH-, -N(O)N-, -CH=CY-, -CH=N(O)-, -C≡C-, -CH₂-CH₂-, -CO-O-, -CH₂O-, -CO-S-, -CH₂-S-, -CH=N-, -COO-Phe-COO- or a single bond, with Y being halogen, preferably chlorine, or -CN.

R' and R'' are, in each case, independently of one another, alkyl, alkenyl, alkoxy, alkenyloxy, alkanoyloxy, alkoxycarbonyl or alkoxycarbonyloxy with 1 to 18, preferably 3 to 12 C atoms, or alternatively one of R' and R'' is -F, -CF₃, -OCF₃, -Cl, -NCS or -CN.

In most of these compounds R' and R'' are, in each case, independently of each another, alkyl, alkenyl or alkoxy with different chain length, wherein the sum of C atoms in nematic media generally is between 2 and 9, preferably between 2 and 7.

5

Many of these compounds or mixtures thereof are commercially available. All of these compounds are either known or can be prepared by methods which are known per se, as described in the literature (for example in the standard works such as Houben-Weyl, Methoden der Organischen Chemie [Methods of Organic Chemistry], Georg-Thieme-Verlag, Stuttgart), to be precise under reaction conditions which are known and suitable for said reactions. Use may also be made here of variants which are known per se, but are not mentioned here.

15

The inventive dopants are in particular useful for anisotropic polymer gels and for cholesteric liquid crystalline materials for cholesteric displays, such as for example phase change displays or polymer free or polymer stabilized cholesteric texture displays (PFCT, PSCT).

20

A further advantage of the chiral dopants according to the invention is that cholesteric liquid crystalline materials comprising these dopants exhibit a low temperature dependence of the reflection wavelength $d\lambda/dT$ (T = temperature, λ = reflection wavelength maximum).

25

Cholesteric displays are described for example in WO 92/19695, WO 93/23496, US 5,453,863 or US 5,493,430, with the entire disclosure of these documents being introduced into this application by way of reference.

30

Furthermore, anisotropic polymer gels and displays comprising them are disclosed for example in DE 195 04 224 and GB 2 279 659.

It has been found that PSCT displays comprising the inventive dopants have reduced response times, lower voltages and improved contrast compared to displays comprising conventional dopants, like

35

e.g. R 811 or CB 15, that are commercially available by Merck KGaA (Darmstadt, Germany). For example, a PSCT display in which a conventional dopant was replaced by a chiral dopant according to the invention showed a reduction of the switching time of up to 50%.

5

A cholesteric film made by using the inventive dopants unexpectedly showed improved brightness, leading to a better contrast between the coloured planar texture and the almost clear focal conic state which is made black using a black backplate.

10

The inventive chiral dopants and liquid crystal materials comprising these dopants are also particularly useful for the preparation of oriented polymer films with a chiral liquid crystalline phase, such as cholesteric or chiral smectic polymer films.

15

Examples of oriented cholesteric polymer films used as broad waveband polarizers can be found in EP 0 606 940, whereas I. Heynderickx and D.J. Broer in Mol.Cryst.Liq.Cryst. 203, 113-126 (1991) describe crosslinked cholesteric polymer films that are made of liquid crystalline diacrylates and contain a low molecular weight chiral dopant. EP 0 562 681 A1 discloses polymer networks with a smectic structure that contains chiral low molar mass compounds and exhibits a piezoelectric effect.

20

25

It has been found that cholesteric polymer films made by using the inventive chiral dopants are brighter compared to films comprising dopants of prior art like e.g. R 811 or CB 15 as mentioned above.

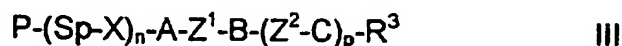
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In order to prepare anisotropic polymer gels or oriented polymer films, the liquid crystalline material in addition to chiral dopants of formula I should also comprise at least one polymerizable mesogenic compound.

35

Thus, another preferred embodiment of the invention are liquid crystalline materials comprising at least one chiral dopant of formula I and at least one polymerizable mesogenic compound.

The polymerizable mesogenic compounds are preferably selected of formula III



5

wherein

10

P is $CH_2=CW-COO-$, $WCH=CH-O-$, $WHC \begin{array}{c} \diagup O \diagdown \\ \text{---} \end{array} CH-$
or $CH_2=CH-Phenyl-(O)_k-$ with W being H, CH_3 or Cl and k being 0 or 1,

Sp is a spacer group having 1 to 20 C atoms,

15

X is a group selected from $-O-$, $-S-$, $-CO-$, $-COO-$, $-OCO-$, $-OCO-O-$, $-S-CO-$, $-CO-S-$ or a single bond,

n is 0 or 1,

20

Z^1 and Z^2 are each independently $-COO-$, $-OCO-$, $-CH_2CH_2-$, $-OCH_2-$, $-CH_2O-$, $-CH=CH-$, $-C\equiv C-$, $-CH=CH-COO-$, $-OCO-CH=CH-$ or a single bond,

25

A, B and C are each independently 1,4-phenylene in which, in addition, one or more CH groups may be replaced by N, 1,4-cyclohexylene in which, in addition, one or two non-adjacent CH_2 groups may be replaced by O and/or S, 1,4-cyclohexenylene, 1,4-bicyclo(2,2,2)octylene, piperidine-1,4-diyl, naphthalene-2,6-diyl, decahydronaphthalene-2,6-diyl, or 1,2,3,4-tetrahydro-naphthalene-2,6-diyl, it being possible for all these groups to be unsubstituted, mono- or polysubstituted with halogen, cyano or nitro groups or alkyl, alkoxy or alkanoyl groups having 1 to 7 C atoms wherein one or more H atoms may be substituted by F or Cl,

30

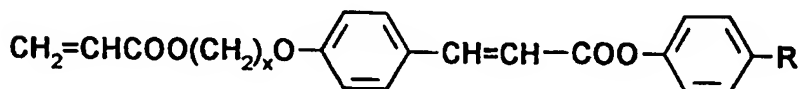
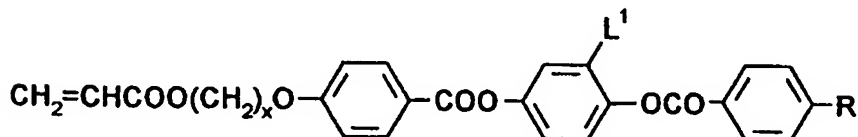
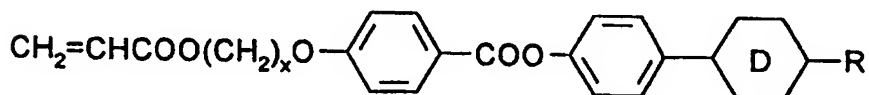
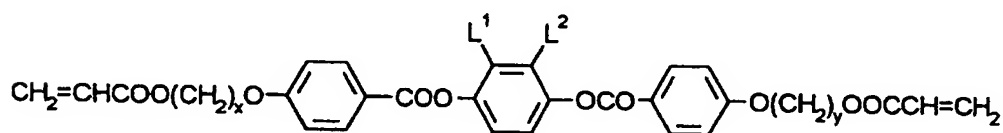
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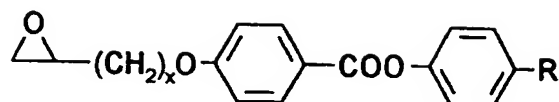
p is 0, 1 or 2, and

R^3 is an alkyl radical with up to 25 C atoms which may be unsubstituted, mono- or polysubstituted by halogen or CN, it being also possible for one or more non-adjacent CH_2 groups to be replaced, in each case independently from one another, by -O-, -S-, -NH-, -N(CH₃)-, -CO-, -COO-, -OCO-, -OCO-O-, -S-CO-, -CO-S- or -C≡C- in such a manner that oxygen atoms are not linked directly to one another, or alternatively R is halogen, cyano or has independently one of the meanings given for P- $(Sp-X)_n$.

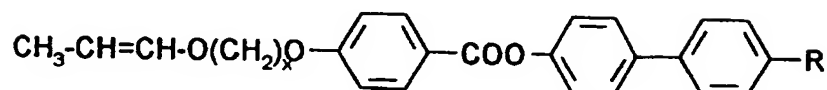
Polymerizable mesogenic compounds according to formula III are described for example in WO 93/22397; EP 0,261,712; DE 195,04,224; DE 4,408,171 or DE 4,405,316. The compounds disclosed in these documents, however, are to be regarded merely as examples that shall not limit the scope of this invention.

Furthermore, typical examples representing polymerizable mesogenic compounds are shown in the following list of compounds, which should, however, be taken only as illustrative and is in no way intended to restrict, but instead to explain the present invention:





5



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wherein x and y are each independently 1 to 12, D is a 1,4-phenylene or 1,4-cyclohexylene group, R is halogen, cyano or an alkyl or alkoxy group with 1 to 12 C atoms and L¹ and L² are each independently H, halogen, CN, or an alkyl, alkoxy or alkanoyl group with 1 to 7 C atoms.

15

The polymerizable mesogenic compounds of formula III can be prepared by methods which are known per se and which are described in the documents cited above and, for example, in standard works of organic chemistry such as, for example, Houben-Weyl, Methoden der organischen Chemie, Thieme-Verlag, Stuttgart.

20

The polymerizable mesogenic compounds of formula III can be mono- or direactive, i.e. they can carry one or two polymerizable functional groups.

25

In a preferred embodiment of the invention the liquid crystalline materials comprise at least one inventive chiral dopant, at least one monoreactive and at least one direactive compound of formula III.

30

In another preferred embodiment the liquid crystalline materials comprise at least one inventive chiral dopant and at least two monoreactive compounds of formula III.

35

Preferably the liquid crystalline materials comprising chiral dopants of formula I and polymerizable mesogenic compounds of formula III additionally comprise a polymerization initiator that decomposes when exposed to actinic radiation, such as for example UV light, or when heated above a certain temperature.

Another object of the invention are polymer films with an oriented chiral liquid crystalline phase obtainable by (co)polymerizing a liquid crystalline material comprising at least one chiral dopant of formula I
5 and at least one polymerizable mesogenic compound preferably selected of formula III.

To prepare polymer films with a chiral liquid crystalline phase with uniform orientation the inventive liquid crystalline materials, for
10 example, are coated onto a substrate, aligned and polymerized in situ by exposing them to heat or actinic radiation. Alignment and curing are carried out in the liquid crystalline phase of the liquid crystalline materials.

15 Polymerization can be achieved for example by using UV light and a photoinitiator that decomposes under UV irradiation. The polymerization may also be started by an initiator that decomposes when heated above a certain temperature.

20 As a substrate for example a glass plate or a plastic film can be used. To achieve uniform alignment, the film for example can be sheared by means of a doctor blade, or shearing can be caused by putting the polymerizable material between two substrates. It is also possible to apply an electric or magnetic field to the coated mixture.
25

A detailed description of the in situ polymerization of polymerizable mesogenic compounds can be found in D.J.Broer et al., Makromolekulare Chemie 190, 2255 (1989).

30 The inventive liquid crystalline materials can additionally comprise one or more other suitable components, such as, for example, catalysts, sensitizers, stabilizers, co-reacting monomers or surface-active compounds.

35 It is also possible to add, for example, non mesogenic compounds with two or more polymerizable functional groups to increase

crosslinking of the polymers. Typical examples for difunctional non mesogenic monomers are alkylidiacrylates or alkylidimethacrylates. Typical examples for non mesogenic monomers with more than two polymerizable groups are trimethylpropanetrimethacrylate or pentaerythritoltetraacrylate.

For the preparation of anisotropic polymer gels, the liquid crystalline materials can be polymerized in situ as described above, however, in this case alignment of the polymerizable material is not necessary.

From the foregoing description, one skilled in the art can easily ascertain the essential characteristics of this invention, and without departing from the spirit and scope thereof, can make various changes and modifications of the invention to adapt it to various usages and conditions.

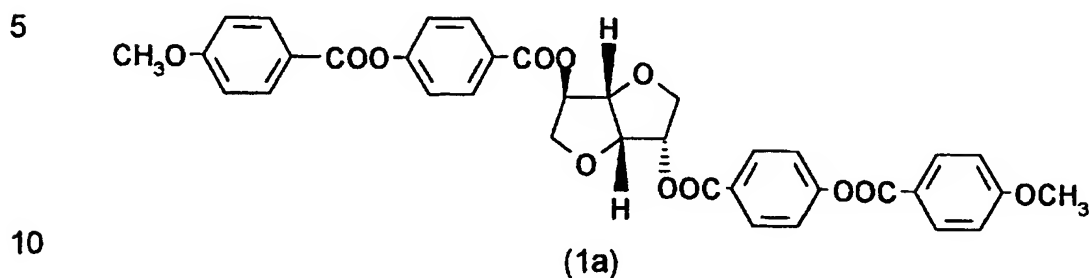
Without further elaboration, it is believed that one skilled in the art can, using the preceding description, utilize the present invention to its fullest extent. The following examples are, therefore, to be construed as merely illustrative and not limitative of the remainder of the disclosure in any way whatsoever.

In the foregoing and in the following examples, unless otherwise indicated, all temperatures are set forth uncorrected in degrees Celsius and all parts and percentages are by weight. The following abbreviations are used to illustrate the liquid crystalline phase behaviour of the compounds:

K = crystalline; N = nematic; S = smectic; Ch = cholesteric; I = isotropic. The numbers between these symbols indicate the phase transition temperatures in degree Celsius.

Example 1

The compound (1a) was prepared according to reaction scheme 1.



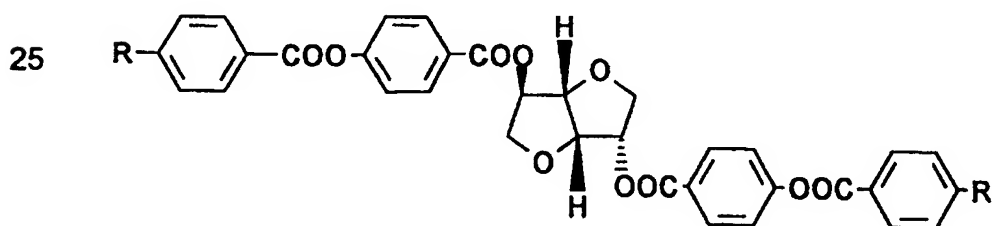
The compound exhibits the phase behaviour K 139 I and shows a very high HTP of $75 \mu\text{m}^{-1}$, determined in the commercially available nematic liquid crystal mixture E 063 (from Merck Ltd., Poole, UK) as a host mixture, which has the following properties

15

clearing point	78.5 °C
birefringence	0.224
dielectric anisotropy	+ 14.6
viscosity (at 20 °C)	38 mm ² /s

20

The following compounds have been prepared analogously



No.	R	phase behaviour	HTP (in E 063)
1b	C ₅ H ₁₁	K 126.1 I	56 μm^{-1}
1c	OC ₈ H ₁₃	K 139.8 I	
1d	C ₇ H ₁₅	K 133 I	

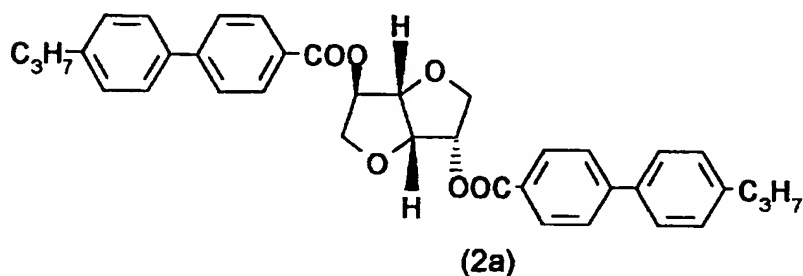
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Example 2

The compound (2a) was prepared according to reaction scheme 2.
The compound has the phase behaviour K 221.9 I.

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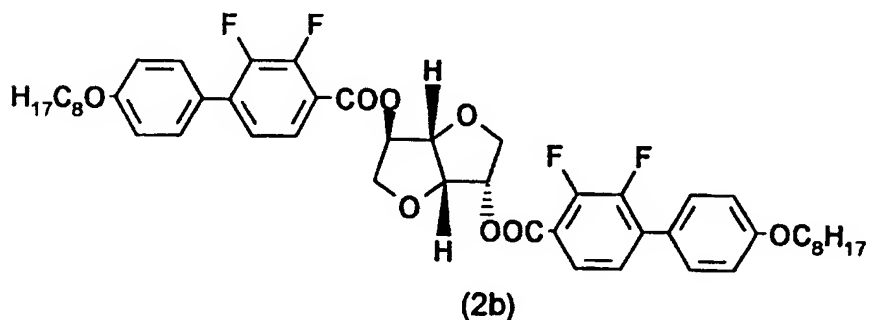
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The compound (2b) was prepared analogously and exhibits the
phase behaviour K 149.4 I.

15

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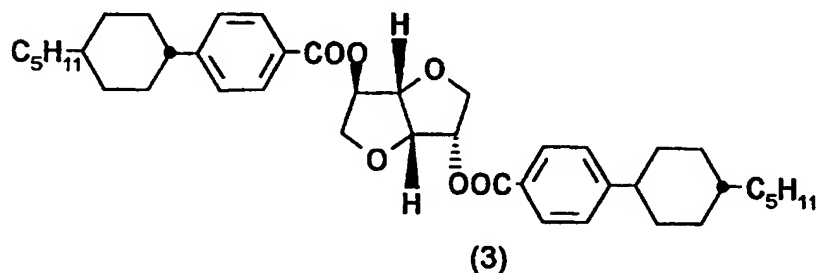
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Example 3

The compound (3) was prepared according to reaction scheme 2.
The compound has the phase behaviour K 177 I and shows a HTP of
36 μm^{-1} measured in the host mixture E 063.

30

35

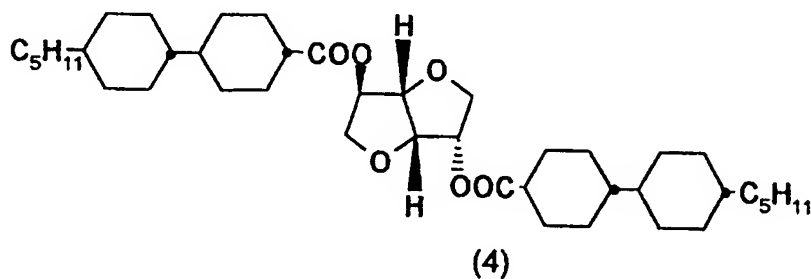


Example 4

The compound (4) was prepared according to reaction scheme 2 and has the phase behaviour K 236 I.

5

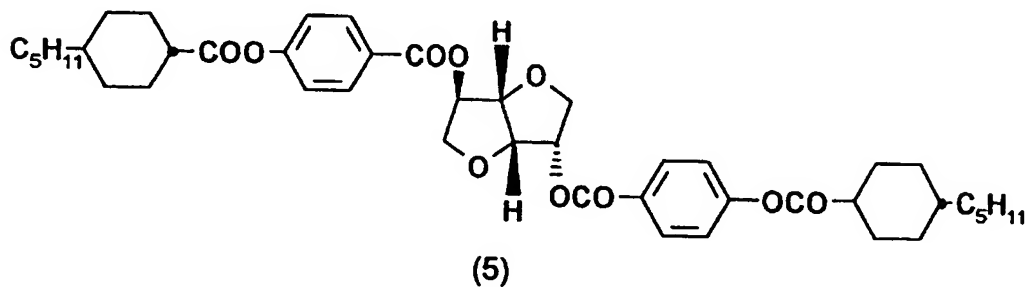
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Example 5

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The compound (5) was prepared according to reaction scheme 1 and has the phase behaviour K 204 I.

20



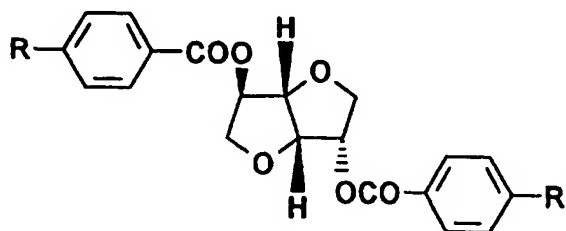
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Example 6

The following compounds were prepared according to reaction scheme 2.

30

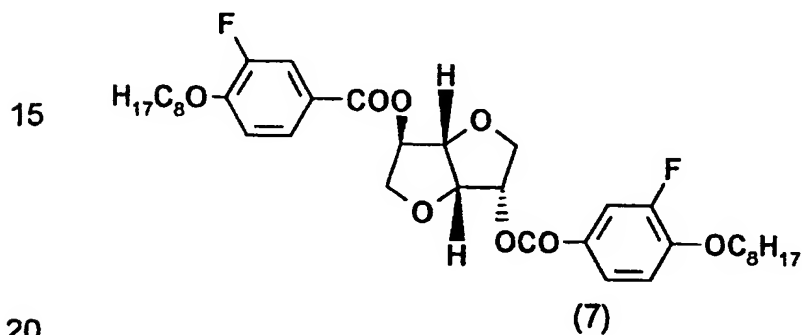
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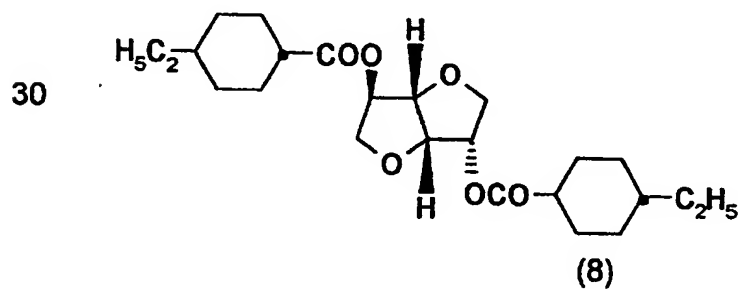
No.	R	phase behaviour	HTP (in E 063)
6a	OCH ₃	K 96.7 I	51 μm^{-1}
6b	OC ₄ H ₉	K 102.3 I	
6c	OC ₆ H ₁₃	K 89.4 I	
6d	C ₅ H ₁₁	K 76.4 I	

Example 7

- 10 The compound (7) was prepared according to reaction scheme 2 and has the phase behaviour K 83.1 I.

Example 8

- 25 The compound (8) was prepared according to reaction scheme 2.
The compound has the phase behaviour K 114 I and shows a HTP of 28 μm^{-1} measured in the host mixture E 063.



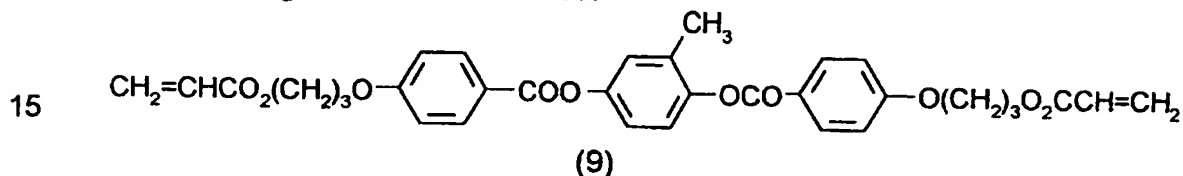
Example 9

The following mixture was formulated

	compound (9)	95.5 %
5	compound (1a)	3.5 %
	Irgacure 369	1.0 %

Irgacure 369 is a commercially available photoinitiator from Ciba Geigy AG (Basel, Switzerland).

Compound (9) is a direactive polymerizable compound which is disclosed e.g. in EP-A2-0 331 233.



The mixture has the phase behaviour Ch 118 I.

20 The mixture was cured as a thin film between two glass substrates using UV light with an irradiance of 3 mW / cm² at 80 °C to give a bright green cholesteric polymer film with a reflection wavelength maximum at 550 nm. The brightness of the film is significantly better than that of a film prepared by using a state of the art chiral dopant
25 such as e.g. R-811 or CB 15 (commercially available from Merck KGaA, Darmstadt).

Example 10

30

Compound (1a) was added at an amount of 3.5 % by weight to the commercially available nematic liquid crystal mixture BL 106 (from Merck Ltd., Poole, UK) exhibiting the following properties:

	clearing point	73.7 °C
35	birefringence	0.213
	diel ctric anisotropy	+ 16.6

After addition of the chiral dopant (1a), the mixture has the phase behaviour Ch 71 I , and shows a reflection wavelength λ of 572 nm. Thus the clearing point of BL106 is only slightly altered by adding the dopant (1a).

5

Example 11

Compound (6b) was added at an amount of 5.5 % by weight to the commercially available nematic liquid crystal mixture E44 (from Merck Ltd, Poole, UK) exhibiting the following properties

clearing point	100 °C
birefringence	0.263
dielectric anisotropy	+ 16.8

15

After addition of the chiral dopant (6b), the mixture has the phase behaviour Ch 88.5 I , and shows a cholesteric reflection wavelength λ of 550 nm at 25 °C and 570 nm at 70 °C, with a low temperature dependence $d\lambda/dT$ of 0.44 nm/ °C.

20

Example 12

Compound (6b) was added at an amount of 5.0 % by weight to the commercially available nematic liquid crystal mixture E63 (from Merck Ltd, Poole, UK) exhibiting the following properties

clearing point	88 °C
birefringence	0.224
dielectric anisotropy	+ 15.6

30

After addition of the chiral dopant (6b), the mixture has the phase behaviour Ch 74.4 I , and shows a cholesteric reflection wavelength λ of 583 nm at 25 °C and 604 nm at 70 °C, with a low temperature dependence $d\lambda/dT$ of 0.47 nm/ °C.

35

Example 13

Compound (6b) was added at an amount of 6.0 % by weight to the commercially available nematic liquid crystal mixture MLC-6422 (from Merck Darmstadt, Germany) exhibiting the following properties

clearing point	93 °C
birefringence	0.1978

After addition of the chiral dopant (6b), the mixture has the phase behaviour Ch 78.5 I , and shows a cholesteric reflection wavelength λ of 550 nm at 25 °C and 606 nm at 70 °C, with a low temperature dependence $d\lambda/dT$ of 1.24 nm/ °C.

Example 14

Compound (6d) was added at an amount of 7.0 % by weight to the nematic liquid crystal mixture MLC-6422.

After addition of the chiral dopant (6d), the mixture has the phase behaviour Ch 72.9 I , and shows a cholesteric reflection wavelength λ of 558 nm at 25 °C and 644 nm at 70 °C, with a low temperature dependence $d\lambda/dT$ of 1.9 nm/ °C.

Example 15

Compound (6b) (7.0 % by weight) and compound (6d) (4.0 % by weight) were added to the nematic liquid crystal mixture MLC-6422.

After addition of the chiral dopants, the mixture shows a cholesteric reflection wavelength λ of 582 nm at 25 °C and 668 nm at 70 °C, with a low temperature dependence $d\lambda/dT$ of 1.9 nm/ °C.

Example 16

- 5 Compound (6c) was added at an amount of 4.5 % by weight to the commercially available nematic liquid crystal mixture BL080 (from Merck Ltd, Poole, UK) exhibiting the following properties
- | | |
|-----------------------|---------|
| clearing point | 87.2 °C |
| birefringence | 0.1546 |
| dielectric anisotropy | + 14.5 |
- 10 After addition of the chiral dopant (6c), the mixture shows a cholesteric reflection wavelength λ of 630 nm at 25 °C and 670 nm at 70 °C, with a low temperature dependence $d\lambda/dT$ of 0.80 nm/ °C.

15 Example 17

- Compound (1a) was added at an amount of 4.5 % by weight to the commercially available nematic liquid crystal mixture BL080. After addition of the chiral dopant (1a), the mixture has the phase
- 20 behaviour Ch 83 I, showing only a slight reduction of the clearing point, and shows a cholesteric reflection wavelength λ of 520 nm at 25 °C and 508 nm at 70 °C, with a low temperature dependence $d\lambda/dT$ of -0.24 nm/ °C.

- 25 The preceding examples can be repeated with similar success by substituting the generically or specifically described reactants and/or operating conditions of this invention for those used in the preceding examples.

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Patent Claims

1. Chiral dopants of formula I

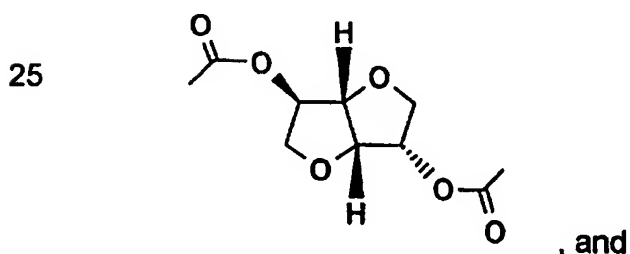


wherein

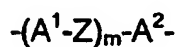
10 R^1 and R^2 are each independently a straight-chain or branched alkyl radical with up to 25 C atoms which may be unsubstituted, mono- or polysubstituted by halogen or CN, it being also possible for one or more non-adjacent CH_2 groups to be replaced, in each case independently from one another, by -O-, -S-, -NH-, -N(CH₃)-, -CO-, -COO-,
15 -OCO-, -OCO-O-, -S-CO-, -CO-S- or -C≡C- in such a manner that oxygen atoms are not linked directly to one another,

20 X^1 and X^2 are independently of each other -O-, -S-, -CO-, -COO-, -OCO-, -OCO-O-, -S-CO-, -CO-S- or a single bond,

G is the following structure element



30 MG^1 and MG^2 are each independently a mesogenic or mesogeneity supporting group of the formula



35 with

Z denoting -COO-, -OCO-, -CH₂CH₂-, -OCH₂-, -CH₂O-,
-CH=CH-, -CH=CH-COO-, -OCO-CH=CH-, -C≡C- or a
single bond,

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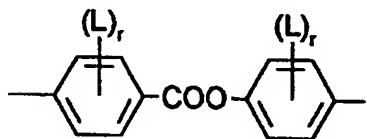
A¹ and A² being each independently 1,4-phenylene in
which, in addition, one or more CH groups may be
replaced by N, 1,4-cyclohexylene in which, in addition,
one or two non-adjacent CH₂ groups may be replaced
by O and/or S, 1,4-cyclohexenylene, 1,4-
bicyclo(2,2,2)octylene, piperidine-1,4-diyl,
naphthalene-2,6-diyl, decahydronaphthalene-2,6-diyl,
or 1,2,3,4-tetrahydronaphthalene-2,6-diyl, it being
possible for all these groups to be unsubstituted,
mono- or polysubstituted with halogen, cyano or nitro
groups or alkyl, alkoxy or alkanoyl groups having 1 to
7 C atoms wherein one or more H atoms may be
substituted by F or Cl, and

20

m being 0, 1, 2 or 3.

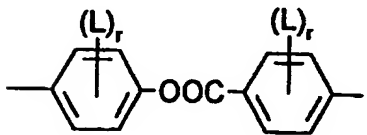
2. Chiral dopants according to claim 1, wherein MG¹ and MG² are
selected from the following formulae

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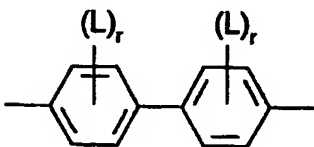
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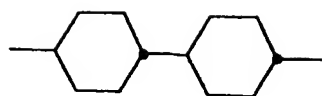


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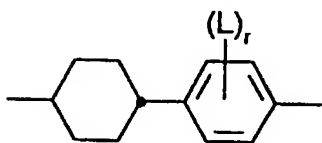


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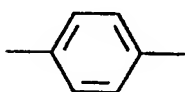
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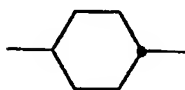
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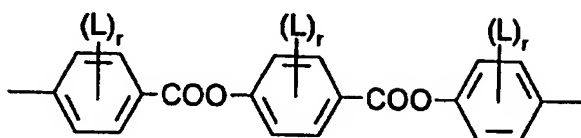
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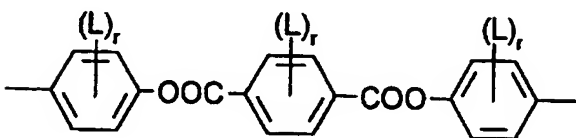
II g

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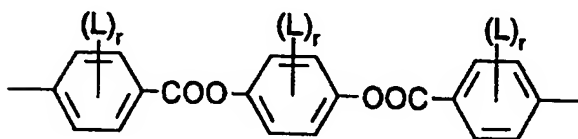
II h

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II i

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II k

wherein L in each case independently denotes halogen, a cyano or nitro group or an alkyl, alkoxy or alkanoyl group having 1 to 7 C atoms wherein one or more H atoms may be substituted by F or Cl and r is 0, 1 or 2.

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3. Chiral dopants according to claim 2, wherein r is 0 and R¹ and R² are each independently alkyl or alkoxy with 1 to 10 C atoms.

4. A liquid crystalline material comprising at least one chiral dopant according to claim 1, 2 or 3.
- 5 5. A liquid crystalline material comprising at least one chiral dopant according to claim 1, 2 or 3 and at least one polymerizable mesogenic compound having at least one polymerizable functional group.
- 10 6. A liquid crystalline material according to claim 4 or 5 comprising 0.001 to 15 % of weight of at least one chiral dopant according to claim 1, 2 or 3.
- 15 7. Use of a liquid crystalline material according to claim 4, 5 or 6 for the preparation of polymer films with a chiral liquid crystalline phase, for active and passive optical elements, colour filters and for liquid crystal displays, for example STN, TN, AMD-TN, temperature compensation, guest-host or phase change displays, or polymer free or polymer stabilized
20 cholesteric texture (PFCT, PSCT) displays.
8. A liquid crystal display comprising a liquid crystalline material according to claim 4, 5 or 6.
- 25 9. A polymer film with a chiral liquid crystalline phase comprising at least one chiral dopant according to claims 1, 2 or 3.
- 30 10. A polymer film with a chiral liquid crystalline phase obtainable by (co)polymerizing a liquid crystalline material according to claim 5 or 6.

INTERNATIONAL SEARCH REPORT

A. CLASSIFICATION OF SUBJECT MATTER IPC 6 C07D493/04 C09K19/58		Int. Application No PCT/EP 97/03167
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC 6 C07D C09K		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practical, search terms used)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	V. VILL ET AL.: "Ferroelectrische Flüssigkristall-Mischungen mit Kohlenhydrat-Derivaten als Dotierstoffe" ZEITSCHRIFT FÜR NATURFORSCHUNG A, JOURNAL OF PHYSICAL SCIENCE, vol. 44a, no. 7, 1989, pages 675-679, XP002042393 see the whole document	1-4,6-8
X	V.VILL ET AL.: "Molekulares Verdrillungsvermögen von Kohlenhydrat-Derivaten" ZEITSCHRIFT FÜR NATURFORSCHUNG A, JOURNAL OF PHYSICAL SCIENCE, vol. 43a, no. 12, 1988, pages 1119-1125, XP002042394 see the whole document	1-4,6-8
-/-		
<input checked="" type="checkbox"/> Further documents are listed in the continuation of box C. <input checked="" type="checkbox"/> Patent family members are listed in annex.		
* Special categories of cited documents : <div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed </div> <div style="width: 45%;"> "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family </div> </div>		
Date of the actual completion of the international search	Date of mailing of the international search report	
2 October 1997	17. 10. 97	
Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer Puetz, C	

INTERNATIONAL SEARCH REPORT

Int. l. Application No
PCT/EP 97/03167

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	DE 43 42 280 A (BASF AG) 14 June 1995 cited in the application see claims; example 2 ---	1-10
A	EP 0 070 401 A (CASSELLA FARBWERKE MAINKUR AG) 26 January 1983 see claims ---	1
A	INDIAN JOURNAL OF CHEMISTRY, SECT. B, vol. 16b, no. 2, 1978, pages 153-155, XP002042395 see compound no. 3 ---	1
P,A	EP 0 747 382 A (BASF AG) 11 December 1996 see claims; examples -----	1-10

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